BIOCHEMISTRY (CH 360, BI 360) COURSE Syllabus and Schedule

FALL SEMESTER 2004

Time Schedule [.]	Lecture: MWF 11:00-11:50 H33		H33		
Time Schedule.	Laboratory Group 1: M 2:00-4:50 pm H		H44		
	Laboratory Group 2:	W	2:00-4:50 pm	H49	
Instructor's name	.	•	•		
address.	G. David Lin		Office: Henry Hall 6		
phone and email.	Phone: 739 8543		e-Mail: dlin@chaminade.edu		
•					
Office Hours:	TR 9:30 – 12:00 am and/or by	arra	ngement.		
	• Sometimes I will be in the classroom 15 minutes before class			efore class and	
	you can ask me questions.				
	• If I have not another commitment after the lecture, I will stay for			I will stay for	
	your questions until	an q	uestions have been answe	ied.	
Course Description:					
	The Biochemistry course (CH 360 or BI 360) enables students to understand biology and life at a chemical level. It provides an overview of general biochemistry with respect to the structure, function, metabolism and mechanism of carbohydrates, lipids, proteins and nucleic acids in biological systems and processes. It also introduces basic techniques in the separation and analysis of these biological molecules, as well as in the characterization of biological processes and their interaction. The course can be used as preparation for specified biochemistry course like medical/clinical/nutritional biochemistry, and analytical/environmental/plant biochemistry, or for related disciplines in forensic science, cellular and molecular biology, pharmacology and toxicology.				
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Course structure	3 x 1-hour lectures per week (3 credits) Plus a 3 hr laboratory work per week (see separate Syllabus CH 360L) (1 credit)				
	Prerequisites : Cellular and Organismic Biology (BI 203/203L and BI 204/204L), and Organic Chemistry (CH 323/323L and CH 324/324L).				

Course Objectives:	On the complete	tion of CH 360,	the students wi	ll be expected t	io:	
	1. Descril	bes sub-cellular	structures of a	living cell with	emphasis on th	neir
	roles ir	the synthesis,	breakdown, trai	nsport and stora	ige of biologica	1
	molecu	iles.		_		
	2. Explain	n the structure,	properties, and	functions of fur	ndamental build	ling
	blocks	(amino acids, c	arbohydrates, li	pids, and nucle	otides), biopoly	ymers
	(nuclei	c acids, peptide	s/ proteins, poly	ysaccharides), r	nembranes, org	anic
	and inc	organic prosthet	ic groups.			
	3. Descril	be the biosynthe	esis and catabol	ism of biologic	al molecules (a	mino
	acids, o	carbohydrates, l	ipids, nucleic a	cids, peptides/p	roteins)	
	4. Explain	n the metabolic	cycles, biologic	cal catalysis and	d kinetics,	
	mechai	nisms, organic a	and inorganic co	ofactors.		66
	5. Discus	s biological equ	illibria and ener	getics with refe	erence to pH/bu	tters,
	binding	g/recognition, p	roton and electr	on transport, or	xidation/reduct	ion,
	macror	nolecular conto	ormations and b	iological interac	ctions that stabi	lize
	biomol	ecules.				
	6. Discus	s biological inic	Drination flow a	nd control thro	ugn DNA replic	cation,
	transcr	ipuon and trans	lation.			
Text Books:	 Required: Horton, R.H.; Moran, L.A.; Ochs, R.S.; Rawn, J.D.; Scrimgeour, K.G. Principles of Biochemistry, 3rd Ed., Prentice-Hall, Inc., Upper Saddle River, NJ, 2002. Supplementary: Holum, J.R. Fundamentals of General, Organic, and Biological Chemistry, 6th edition, John Wiley and Sons, Inc. Hoboken, NJ, 1998. Lehninger, A.L Principles of Biochemistry, 4th ed. Worth Publisher, New York.2005. Gilbert, H.F. Basic Concepts in Biochemistry: a student's survival guide. McGraw-Hill, New York.1992. 					
Scale:	learned. Your	lowest exam c	of the three ter	m exams will	be counted on	ly half
	as much as the	e other exams,	but the final e	xam will not t	be affected. Th	ne
	grading scale	will be based of	on the followir	ng:		
	00.100.0/	00.00.01		7 0 <i>6</i> 0 <i>0</i> /	0.70.04	7
	90-100 %	80-89 %	/0-/9 %	50-69 %	0-50 %	-
	Α	В	С	D	F	
	Missing an ex- exam is given scheduling con- the exam. In the notify there In- doctors note of receive full creations	am can have so to try to make nflict you mus ne case of an u structor before r similar evide edit for a make	erious conseque o up an exam. I t make arrange inexpected illn e the exam is g ence to provide e-up exam.	tences. Do not if you know the ements with the less or problem given and be propriet a valid excus	t wait until aft hat you have a he Instructor p n you must sti repared to pre se. You might	er the rior to ll sent a not

CH360 Tentative Lecture Schedules

Week	Date	Chapter	Contents
1	0/22	in Text	
	8/23	1.0	Welcome Pre-test.
1	8/25	1, 2	Pre-test feedback Biochemistry, Organic Chemistry, General
	l		Chemistry, and their links Living organisms, dynamic processes,
	l		cells, biological molecules, biopolymers, and their interactions
1	0/07	<u> </u>	The amazing water.
	8/21	2	No Class: Spiritual Convocation,
2	8/30	3	Structures and properties of amino acids Pepudes, primary
	l		structures of proteins, protein purfilication and sequencing
2	0/1		Secondary tertiary and quaternary structures of proteins Protein
۷.	9/1	4	folding and stability
2	0/3		Clobular fibrous and membrane proteins Hemoglobin
<i></i>	7/5		myoglobin collagen antibodies Biological function of proteins
3	9/6	+	No class: Labor Day
3	9/8	5	Properties of enzymes: Classification and nomenclature Catalytic
5	2/0	5	efficiency and specificity Kinetics Inhibition
3	9/10	6	Mechanisms of enzyme: acid-base catalysis covalent catalysis
5	<i>J</i> /10		proximity and transition state. Key-lock model, induced fit
	l		concept, kinase, lysozyme, and proteases.
4	9/13	7	Enzyme cofactors: Essential inorganic ions (K, Mg, Ca, Zn, Fe),
		-	Group-transferring coenzymes (NAD ⁺ , NADP ⁺ , FAD, FMN).
	l		Review so far
4	9/15	1	FIRST MIDTERM EXAM
4	9/17	7	Group-transferring coenzymes (CoA, vitamins, biotin, ubiquinone,
	l		lipoamide, cytochromes and some proteins). Regulation of
			enzyme
5	9/20	8	Carbohydrates overview Aminosaccharides Glycans,
	l		proteoglycans, cartilage Peptidoglycans, cell wall, penicillin
	<u> </u>		mode of action Glycoproteins
5	9/22	9	Structure and property of lipids: Fatty acids, triacylglycerols (fats
	l		and oils), glycerophospholipids (in cell membrane), sphingolipids,
			eicosanoids, steroids, and terpenes, waxes.
5	9/24	9	Cell membrane: lipid bilayers, membrane proteins, active and
	l		passive transport, signal transduction (G proteins and protein
	0/07	10	kinases)
6	9/27	10	Metabolism: Metabolic pathways, catabolic and anabolic reactions,
	0/20		thermodynamics and free energy changes, Energy-rich molecules
6	9/29		Glycosis: A ubiquitous ten step pathway, a 4-step regulation
6	10/1	12	The citric acid cycle: The cross road in the aerobic catabolism of
	l		carbohydrates, lipids and amino acids. The eight enzyme-catalyzed
	1		reaction

7	10/4	13	Additional Pathways in Carbohydrate Metabolism: Glycogen
/	10/ 4	15	degradation: Gluconeogenesis Pentose phosphate pathway Inter-
			conversions between a ketose and an aldose phosphate
7	10/6	14	Ovidetive phosphorylation: The mitochondrion, the respiratory
/	10/0	14	observe phospholylation. The initochondrion, the respiratory
			electron transport chain, the proton concentration gradient, the
	10/0	1.4	coupling
/	10/8	14	The chemiosmotic theory, the protonmotive force
8	10/11		No Class: Discoverer's Day
8	10/13	14	Electrochemistry of coenzymes, the P/O ratio, uncouplers,
			regulation of oxidative phosphorylation
8	10/15	15	Photosynthesis: Chloroplasts, chlorophylls, the light reaction, the
			dark reaction, CO ₂ fixation by the RPP Cycle, Comparisons to
			oxidative phosphorylation
9	10/18	16	Lipid Metabolism: Absorption, storage and degradation of fatty
			acid, β -Oxidation of odd-chain and unsaturated fatty acids
9	10/20		SECOND MIDTERM EXAM
9	10/22	16	Lipid metabolism: Fatty acid Synthesis, cholesterol synthesis
10	10/25	16	Lipid Metabolism: Eicosanoids, triacylglycerols,
			glycerophospholipids
10	10/27	17	Nitrogen metabolism: sources of nitrogen
10	10/29	17	Nitrogen metabolism: synthesis of amino acids
11	11/1	17	Nitrogen metabolism: degradation of amino acids
11	11/3	18	Nitrogen metabolism: Nucleotide, alkaloids
11	11/5	19	Nucleic acids: Nucleotides, DNA, the double helical structure.
	11/0		RNA
12	11/8	19	Nucleic acids: Histories, chromatins, chromosomes, hydrolysis of
	11/0	17	DNA by restriction endonucleases
12	11/10	20	DNA replication: The central dogma DNA polymerases the
12	11,10	20	replicsome the replication fork
12	11/12	20	DNA replication: regulations DNA sequencing DNA repair and
12	11/12	20	recombination
13	11/15	21	Transcription: DNA-directed RNA synthesis basic steps
13	11/17	21	Transcription: regulation and RNA processing:
13	11/17	<i>4</i> 1	THIRD MIDTERM EXAM
13	11/19	$\gamma\gamma$	Protein synthesis: Codons tRNA aminoacyl tRNA synthesis
14	11/22		ribosomes
14	11/24	$\gamma\gamma$	Translation: From mPNA to protein basic stans and regulations
14	11/24		No Close: Thanksgiving Prook
14	11/20	22	Pacombinant DNA Tachnology: The basic stone, DCD amplication
13	11/29	23	of specific DNA site directed mutagenesis
15	10/1		Deview
15	$\frac{12/1}{12/2}$		NEVIEW Dest test
15	12/3		Post-test
	10/2.0		
	12/6-9		FINAL EXAM (specific date and time to be advised)