

PM FE

C4 103L30

Course Title: College Chemistry Lab **103L**  
Term: Fall 2000  
Time: Saturday 800 - 1210  
Location: **Chaminade** Main Campus Henry Hall  
Instructor: Ada Tomosada

Lab Manual:

There is no lab **manual**. **Handouts** will be provided by the **instructor**.

Objectives:

**Together** with Chemistry 103 lecture section, the lab section is designed to enhance your understanding of scientific methods and concepts. **Experimental** work brings a practical understanding of chemistry and hands-on experience in **different** techniques.

Safety Requirements:

Students are required to practice **safety** precautions such **as wearing** *safety* glasses while performing experiments. Also covered shoes **are required**, and long pants is recommended. Hair **must** be tied back away **from the face**. It is suggested **that** the **student wear** very **casual attire** since clothing is easily soiled during **laboratory** work.

**Only** registered **students** will be allowed in **the** laboratory.

Grading:

There **will** be ten **experiments** performed and **therefore** ten **lab** reports to be handed in. (Lab **reports** are to be handed in **ON the following lab meeting.**) A quiz will be given at the **beginning** of each lab session **starting** on October 14. (the second **lab meeting**) **This quiz will** cover the **experiment done** the previous week. A final exam will be given on the **last lab** meeting **covering** all **material**.

Grading breakdown is as follows: **40%** lab participation and reports, **25%** lab quizzes, **15%** attitude (following **safety requirements**, etc.), **20%** **final** exam. Make-up labs will be offered with valid excuse.

Tentative Lab Schedule:

|                       |                                     |
|-----------------------|-------------------------------------|
| Week 1 (10/7)         | Introduction to <b>Measurements</b> |
| Week 2 (10/14)        | A Graphic Experience                |
| Week 3 (10/21)        | Significant Figures                 |
| Week 4 (10/28)        | Empirical Formula                   |
| Week 5 (11/4)         | Stoichiometry                       |
| Week 6 (11/11)        | Conservation of Matter              |
| Week 7 (11/18)        | <b>Spectrophotometry</b>            |
| Week 8 (11/25)        | no lab (Thanksgiving)               |
| Week 9 (12/2)         | Acid - Base Titration               |
| <b>Week</b> 10 (12/9) | <b>Final</b> Exam                   |

Course Title: College **Chemistry** 103

Term: Fall 2000

Location: Tripler Hospital

Time: 17:30 - 19:35

Instructor: Ada Tomosada

Communications: telephone 7349424  
pager 680 2802 (Phone # 10360)

Textbook: **Principles and Applications** of Inorganic, **Organic** & Biological **Chemistry**  
Caret, Denniston, and Topping, 1997

Course Description:

College Chemistry 103 is an **introductory** course for students who may be interested in continuing their **education** in the sciences or other technological fields. The course lecture along with laboratory work is a step by step procedure in **introducing** science methods and concepts to the student who has little or no chemistry background.

Course Objectives:

This course is designed to **familiarize** you with the **concepts** of chemistry that may be used as a basis for other more **intensive** courses in the science field.

The course will cover the **first** ten chapters of the textbook.

Course Requirements:

**Scientific** calculator is required. Concurrent enrollment in Chemistry 103L, and high school algebra is recommended.

Only **registered** students will be allowed to attend **classes**.

Grading:

A quiz will be given after each chapter. (**Usually** on Wednesday) Homework assignments will be given, and should be done in preparation for quizzes. A final **exam** will be given on the last day of class. Calculators are allowed during quizzes and **final** exam. All work must be shown on paper for quizzes and **final** exam.

Grading breakdown is as follows: 55% quizzes, 35% **final** exam, 10% attendance and attitude. Letter **grades** will be assigned according to a class curve.

Tentative Class **Timetable:**

|         |  |
|---------|--|
| Week 1  | <b>Chapter</b> 1 Chemistry Methods and <b>Measurements</b><br>Chapter 2 The <b>Structure</b> of the Atom |
| Week 2  | Chapter 2 The <b>Structure</b> of the Atom (quiz Chapt 1)  |
| Week 3  | Chapter 3 Elements, Atoms and the Periodic Table (quiz Chapt 2)  |
| Week 4  | Chapter 4 Structure and Properties of Ionic and Covalent<br>Compounds (quiz Chapt 3)                     |
| Week 5  | Chapter 5 Calculations <b>and the</b> Chemical Equation (quiz Chapt 4)                                   |
| Week 6  | Chapter 6 States of Matter (quiz Chapt 5)  |
| Week 7  | <b>Chapter</b> 7 Reactions and Solutions (quiz Chapt 6)  |
| Week 8  | Chapter 8 Chemical and Physical Change (quiz Chapt 7)  |
| Week 9  | <b>Chapter</b> 9 Charge-Transfer Reactions (quiz Chapt 8)  |
| Week 10 | Chapter 10 Radioactivity and Nuclear Medicine (quiz Chapt 9)   |
| Week 11 | Organic Compounds and Review for <b>final</b> exam (quiz Chapt 10)<br><b>Final Exam</b>                  |