BIOCHEMISTRY (CH 360) COURSE OUTLINE FALL SEMESTER 2001

Bülent Terem

Required Textbook:	Horton, R.H.; Moran, L.A.; Ochs, R.S.; Rawn, J.D.; Scrimgeour, K.G. <i>Principles</i> of <i>Biochemistry</i> , 3 [°] Ed., Prentice-Hall, Inc., Upper Saddle River, NJ, 2002.		
Supplementary Books		v, 4 [°] h Ed,; W.H. Freeman, 1995.	
	Carey, Francis A. Orga	anic Chemistry, McGraw-Hill, 4'h Ed., 2000 Chemistry, Prentice-Hall, Inc., 1995.	
Exams and Grading:	Three 45-minute midterms, quizzes and homework assignments, and a ninety minute comprehensive final.		
	Course Grade (all grades in percentages) - + .26 (average of the three midterms) + .26 (average of the two highest midterms) + .10 (average of the quizzes and assignments) +.38 (final)		
Office Hours:	MWF 10:00 - 11:00 pm and/or by arrangement		
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Course Objectives:

This biochemistry course is aimed at students with a solid background in organic chemistry and an interest in molecular aspects of biological processes. Amongst the medley of topics to be discussed three different perspectives are adopted: The course starts with a discussion of familiar topics and principles in a qualitative fashion, emphasizing their applicability to biological systems. What follows is a more descriptive study of well-known metabolic pathways while questioning all possible variations in an attempt to understand each mechanism. The final perspective is in line with the "holistic" nature of biochemistry, where the links and relationships of individual **metabolic pathways** with one another are elucidated when biosynthetic origins of well-known biological metabolites are investigated. During the semester brief journeys into molecular biology will help to illustrate the enormous potential of the field. Further principles of nucleic acid chemistry will be introduced during the lab course.

Upon completing the course the students should be ready for more specialized courses in clinical biochemistry and molecular biology.

Week	Date	Chapter in Text	Subject
1	8/27	1	Introduction to Biochemistry Functional groups; Linking metabolites Biological Polymers Laws of <u>Thermodynamics</u> (revisited)
1	8/29	Ι	Cells Living organisms, and evolution
1	8/31	2	Water and non-covalent interactions Water as a nucleo hile: Hydrolysis
2	9/3		No class: Labor Day
2	9/5	2	Ionization of water Buffers
		3	Amino Acids and the primary structures of proteins Ionization of amino acids
2	9/7	3 4	Protein purification techniques Protein sequencing Protein Synthesis (review) Secondary Structure of Proteins <u>Tertiary</u> Structures
3	9/10	4	Globular proteins Oxygen bindin : M o lobin and Hemoglobin
3	9/12	5	Enzymes: Classification Enzyme kinetics: The Michaelis-Menten Equation
3	9/12	5	Reversible and irreversible <u>enzyme</u> inhibition
4	9/17	5	Allosteric <u>enzymes</u> Review
4	9/19	5	FIRST MIDTERM EXAM
4	9/19		
4 5	9/31 9/24	6 b	General features of fine mechanisms Chemical <u>Catalysis Proximity</u> Effect Transition state stabilization TS analogs as inhibitors
5	9/36	6	Mechanism of Serine protease activity
5	9/28	7	Coenzymes: Classifications; NAD", NADH, FAD, FMN, CoA,
6	10/1	7	TPP, idoxal Phosphate, Biotin, Tetrah drofolate, Cobalamin, Li oamide,
6	10/3	7	Lipid vitamins, Ubi uinone, C ochromes
6	10/5	8	Carbohydrates (review) Aminosaccharides
7	1013	0	No Class: Discoverer's <u>Day</u>
7	10/10	8	Pe tido cans Nucleosides (review) Penicillin mode of action
	10/10		
7 8		9 9	Kinase <u>Catalysis</u> <u>Lipids: Fatty</u> acid derivatives
8	10/15	9	Nucleotide derivatives Nucleotides as regulatory molecules Phospholipids; S hingoli ids; Steroids Lipid bila ers
8	10/17	9	Membrane proteins Membrane Transport
8	10/19	9	Transduction of extracellular signals
9	10/22	9	Review
9	0/24		SECOND MIDTERM EXAM
9	0/26	10	Metabolism: As sum of cellular reactions; as sequence of stepwise reactions; as regulated <u>pathways</u> Catabolic <u>pathways</u> "High <u>Energy</u> " metabolites
10	0/29	10	Free <u>energy</u> change and reduction <u>potential: Electrochemistry</u> of <u>coenzymes</u>
10	0/31	11	Glycol sis: Mechanism of individual steps
10	11/2	11	Thermodynamics and Regulation of gylcolysis
	11/2	12	Citric Acid Cycle: Individual steps, Regulation of the cycle
11 11	11/3	12	En and exit metabolites Gl ox late <u>Cycle</u>
		12	Additional <u>Carbohydrate Pathways: Glycogen</u> Degradation; Gluconeo enesis
11	11/9	15	
12	<u>11/12</u>	1.2	No Class: Veteran's <u>Day</u>
13	11/14	13	Variations in Gluconeogenesis The Pentose Phosphate Pathway Interconversions
10	11/1-	14	catalyzed b Transketolase and Transaldolase
13	11/16	14	Oxidative Phosphorylation: The chemiosmotic theory; The protonmotive force;
		14	<u>Thermodynamics</u> of electron transport Cofactors and Complexes in electron transport Active transport of ATP, ADP, and
13	1 1119	14	Pi across the Mitochondrial Membrane The P:O Ratio
13	11/21	14 15	Uncouplers Regulation of Oxidative Phosphorylation Photosynthesis: Light reactions and dark reactions Comparisons to oxidative
			hos ho lation
13	11/23		No Class: Thanksgiving Break
14	1/26		THIRD MIDTERM EXAM
14	11/28	15	The RPP <u>Cycle</u> CO, Fixation
14	11/30	16	Lipid Metabolism: Fatty acid oxidation (3-Oxidation of odd-chain and unsaturated fatty acids
15	12/3	16	Eatty acid Synthesis
15	12/5	16	Biosynthesis of other related metabolites
15	12/7	10	Review
	12/10		FINAL EXAM (Monday; 8:00).