## ORGANIC CHEMISTRY II (CH 324) COURSE OUTLINE SPRING SEMESTER 1999

## Bülent Terem

Required Textbook:	Brown, W.H. Organic	Chemistry, Saunders College Publishing, 1995.	
Supplementary Books	Iverson, B.L. and Iverse <i>Chemistry by William F</i> 1995.	on, S.A. Student Guide and Problems Book for Organic H. Brown, Vol. 1 and Vol. 2, Saunders College Publishing, hemistry, Prentice-Hall, Inc., 1995.	
Software:	ChemDraw (for Windo	ws); Chem 3-D, Cambridge Soft Corp.	
Supplementary Mater	ials: Molecular Models		
Exams and Grading:	Three 70-minute midterms, quizzes and homework assignments, and a ninety minute comprehensive final.		
	+ .26 (average	es in percentages) of the three midterms) of the two highest midterms) of the quizzes and assignments)	
Office Hours:	TR 12:30 -1:00 and/or	by arrangement	
Review Sessions:	Saturday mornings		
Office:	Henry Hall 45 e-mail:	(Ph: 735-4806) terem@gold.chem.hawaii.edu	

General Comments:

During Chem 323 the focus was on understanding the basic concepts of organic reaction mechanisms. With a sound grasp of introductory principles, it will be much easier, and more fulfilling, to follow the rationale of the second semester's organic chemistry course. In the next fifteen weeks we will start with topics in physical organic chemistry and gradually move into biological organic chemistry. The chemistry of aromatic compounds will provide a framework where structure-reactivity relationships are clearly laid out. An introduction to spectroscopic techniques will show how physical methods can illustrate the properties of organic compounds. As we go into the chemistry of carbonyl compounds, and sample in-depth analyses of certain reactions, comparisons with biological pathways will stand out. Meanwhile, multi-step syntheses will elucidate common logical strategies. Finally, the chemistry of carbohydrates, amino acids and proteins will give the course a biological flavor.

As far as study habits are concerned, any effort to understand the material as it is presented, or shortly after, is the key to success in this course. Overlaps between lab and lecture, as well as review sessions, are designed to emphasize this point. A "diligent" student is also expected to attempt most of the problems in the relevant sections of the textbook. It is hoped that at the end of the spring semester, each student will possess a sound molecular background to help him/her interpret what goes on in biological systems.

	ı ı		1
r	е <sup>к</sup> у		
	.Z .		· · · · · · · · · · · · · · · · · · ·
1	1/12	12	A general look into spectroscopy: an organic chemist's perspective Mass
			spectrometry Instrumental features Analysis of mass spectra: Resolution;
			Isotopes; Molecular ion peaks
1	1/14	12	Fragmentation in mass spectrometry Stable fragments Rearrangements
		15	Structure determination using mass spectral data
			A review of organic reactions Benzene derivatives and aromatic compounds The
			concept of aromatici
2	1/19	15	Annulenes Heterocyclic aromatic compounds Aromatic ions Fused ring
			aromatic compounds Phenols Side chain reactions of aromatic compounds
	1/01	1.6	Mass <u>spectrometry</u> of benzene derivatives
2	1/21	16	General mechanism of electrophilic aromatic substitution reactions Halogenation;
	1/07	16	Nitration; Sulfonation
3	1/26	16	Friedel Craft's alkylation and acylation reactions Substituent effects in
2	1/20	16	electro hilic aromatic substitution Substituent effects in electro hilic aromatic substitution cont
3	1/28	16	
4 4	2/2 2/4	13	<b>FIRST</b> MIDTERM EXAM Nuclear Magnetic Resonance (NMR) Spectroscopy: The origin of NMR absorptions;
4	2/4	15	'H NMR
5	2/9	13	Equivalence in NMR Spectroscopy Integration Chemical shifts
5	2/11	13	Sin-sin <u>coupling</u> C NMR Interpretation of NMR spectra
6	2/16	13	Magnetic induction in a $\pi$ -system NMR Spectroscopy of benzene derivatives
-	_, _ *		15.4B Structure determination using spectroscopic data
6	2/18	17	Aldehydes and ketones: Classification and their spectroscopic properties
			Preparation
7	2/23	17	Addition of nucleophiles to aldehydes and ketones: General mechanism (acid and
			base catalyzed) Carbon nucleophiles: Grignard reagents; Organolithium reagents;
			Ace lides; Cyanide
7	2/25	17	The Wittig Reaction' Oxygen Nucleophiles: Hydrates; Acetals Acetals as
			protecting groups Thioacetals 1,3-dithiane anions
8	3/2	17	Nitrogen nucleophiles Oxidation of aldehydes and ketones Baeyer-Villiger
			oxidation Reduction of <b>aldeh des</b> and ketones
8	3/4	17	Keto-enol tautomerism Reactions at the a-carbon: Racemization; H-D exchange;
		;	Aldol Reaction
9	<u>3/9</u>		SECOND MIDTERM EXAM

	1 1		
9	3/11	18	Carbohydrates: Structural classifications; <b>hemiacetal</b> formation (cyclization) of monosaccharides Conformations of <b>monosaccharides</b> Reactions of monosaccharides: oxidation; reduction; acetal formation N-glycosides:
10	3/16	18	Nucleosides, nucleotides; linkages in nucleic acids (25.1-25.2)         Fischer's roof of glucose structure       Disaccharides
10	3/18	18 19	Biologically active carbohydrates Polysaccharides Carboxylic acids: Structural features; Effects of substituents on <u>acidity</u>
11	3/22 -3/26		SPRING BREAK
12	3/30	19	Preparation of carboxylic acids: Hydrolysis of nitriles; Carbonation of Grignard rea ents; Oxidation reactions
12	4/1	19 20	<ul> <li>Reactions of carboxylic acids: Reduction; Esterification; Thermal decarboxylation of β-ketoacids</li> <li>Carboxylic acid functional derivatives: Esters, Amides, Anhydrides, Acid halides,</li> </ul>
12	4/6	20	Nitriles Reactions of carboxylic acid derivatives: Mechanism of nucleophilic acyl
13	4/0	20	substitution Hydrolysis Comparison of the reactivities of carboxylic acid derivatives Interconversion of <u>carboxylic</u> acid derivatives
13	4/8	20	The Hofinann Rearrangement Condensation of Esters: Claisen Condensation; Hydrolysis and decarboxylation of $\beta$ -ketoesters Carboxylic acid derivatives of <b>biolo ical</b> interest Sulfonamides
14	4/13	21	Enolate anions: directed <b>aldol</b> reactions Acetoacetic and <b>malonic</b> ester syntheses a, -unsaturated <b>aldeh des</b> and <b>ketones</b> The Michael reaction
14	4/15	21	Enamine reactions Robinson Annulation Reaction Review
15	4/20		THIRD MIDTERM EXAM
15	4/22	22	Amines: Nomenclature; structure and bonding; basicity Syntheses of amines: Gabriel Synthesis; Reductive routes; Hofinann (and Curtius) Rearrangements Reactions of amines: Hinsber Test; Hofinann Elimination
16	4/27	22	lamines: Nitrosation reactions 1 diazonium salts: Sandme er reactions
16	4/29	24	<ul> <li>Amino acids: Structural-stereochemical features; acid-base properties Peptides and proteins Peptide structure determination: End group analyses; partial hydrolyses</li> <li>S thetic strategies in peptide syntheses</li> </ul>
	515		FINAL EXAM (Wednesday; 08:00)